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D 8000 41 - 012 High Li<sup>+</sup>-Ion Storage Capacity and Double-Electrochromic Behavior of Sol-Gel-Derived Iron Oxide Thin Films with Sulfate Residues. - Iron oxide films are fabricated by spin-coating sols (obtained from FeSO<sub>4</sub> by precipitation using  $NH_4OH$  solutions in the presence of  $H_2O_2$  as an oxidant) onto ITO-coated glass substrates followed by annealing at 350 °C or above. The samples are characterized by TG-DTA, FTIR, XRD, TEM, and XPS. The results show that sulfate residues are adsorbed on the surface of  $\mathrm{Fe_2O_3}$ nanoparticles. Cyclic voltammetry reveals that the films annealed at 350 °C exhibit a Li<sup>+</sup>-ion storage capacity as high as ca.  $0.50 \text{ mC/cm}^2$  per nanometer of the film thickness and show double-electrochromic behavior depending on the extent of intercalation during the electrochemical lithium insertion/extraction processes. The high charge capacity and nearly optical passiveness of the  $\mathrm{Fe_2O_3}$ films suggest their promising applications as counter-electrodes in electrochromic devices. — (WANG, ZHONGCHUN; HU, XINGFANG; KAELL, PER-OLOV; HELMERSSON, ULF; Chem. Mater. 13 (2001) 6, 1976-1983; Dep. Eng. Quim., Inst. Super. Tec., Tech. Univ. Lisbon, P-1096 Lisboa, Port.; EN)

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